## **Lesson Plan**

Name of the Faculty : Sh. Laxman Ram

Discipline : Mechanical Engineering

Semester : 3rd

Subject : Thermodynamics-I

Lesson Plan duration : 15 weeks (01.09.2023 to 15.12.2023)

Work load per week : Lecture -03, Practical -02

Week	Theory		EXECUTION	
	Lecture	Topic (Including assessment/tast)	Date	Sign.
1 <sup>st</sup>	Day 1 <sup>st</sup>	(Including assessment/test) Introduction about the subject & brief overview.		
	_	January Control of the Control of th		
	2 <sup>nd</sup>	Unit I:		
		1. Fundamental Concepts:		
		Thermodynamic state and system, boundary, surrounding, universe,		
	3 <sup>rd</sup>	thermodynamic systems – closed, open, isolated, adiabatic,		
		homogeneous and heterogeneous, macroscopic and		
		microscopic		
$2^{\text{nd}}$	4 <sup>th</sup>	Properties of system – intensive and extensive,		
		thermodynamic equilibrium, Quasi – static process, reversible		
	4h	and irreversible processes,		
	5 <sup>th</sup>	Zeroth law of thermodynamics, definition of properties like		
		pressure, volume, temperature, enthalpy and internal energy.		
	6 <sup>th</sup>	2. Laws of Perfect Gases:		
		Definition of gases, explanation of perfect gas laws – Boyle's		
		law, Charle's law,		
$3^{\rm rd}$	7 <sup>th</sup>	Avagadro's law, Regnault's law, Universal gas constant,		
	41-	Characteristic gas constants and its derivation.		
	8 <sup>th</sup>	Specific heat at constant pressure, specific heat at constant		
	- th	volume of a gas,		
	9 <sup>th</sup>	derivation of an expression for specific heats with		
4 <sup>th</sup>	10 <sup>th</sup>	characteristics,		
4	10 11 <sup>th</sup>	simple numerical problems on gas eqn.		
	11	Unit II:		
		3. Thermodynamic Processes:  Types of thermodynamic processes. Isochoric process		
		Types of thermodynamic processes – Isochoric process, equation representing the process. Derivation of work done,		
		change in internal energy, change in entropy, rate of heat		
		transfer for the above process.		
	12 <sup>th</sup>	Isobaric process, equation representing the process.		
		Derivation of work done, change in internal energy, change in		
		entropy, rate of heat transfer for the above process.		
5 <sup>th</sup>	13 <sup>th</sup>	Isothermal process, equation representing the process.		

		Desiration of and described in internal account in	
		Derivation of work done, change in internal energy, change in	
	14 <sup>th</sup>	entropy, rate of heat transfer for the above process.	
	14**	Adiabatic, isentropic processes, equations representing the	
		processes.	
		Derivation of work done, change in internal energy, change in	
	₄ <b>≂</b> th	entropy, rate of heat transfer for the above processes.	
	15 <sup>th</sup>	Polytropic and throttling processes, equations representing the	
		processes.	
		Derivation of work done, change in internal energy, change in	
-th	, -th	entropy, rate of heat transfer for the above processes.	
6 <sup>th</sup>	16 <sup>th</sup>	1 <sup>st</sup> sessional test (Tentative)	
	41-		
	17 <sup>th</sup>	Assessment	
	18 <sup>th</sup>	Unit III	
		4. Laws of Thermodynamics	
		Laws of conservation of energy, first law of thermodynamics	
		(Joule's experiment) and its limitations,	
$7^{\mathrm{th}}$	19 <sup>th</sup>	Steady flow energy equation, Application of steady flow	<u> </u>
		energy equation for turbines, pump, boilers, compressors,	
		nozzles, and evaporators.	
	$20^{\text{th}}$	Heat source and sink, statements of second laws of	
		thermodynamics: Kelvin Planck's statement, Classius	
		statement,	
	21 <sup>st</sup>	Equivalency of statements, Perpetual motion Machine of first	
		kind, second kind, Carnot engine,	
$8^{th}$	$22^{\rm nd}$	Introduction of third law of thermodynamics, concept of	
		irreversibility and concept of entropy.	
	23 <sup>rd</sup>	Unit IV	
		5. Steam Generators:	
		Uses of steam, classification of boilers, Comparison of fire	
		tube and water tube boilers.	
	24 <sup>th</sup>	Construction and working of Nestler boiler, Babcock &	
		Wilcox Boiler.	
9 <sup>th</sup>	25 <sup>th</sup>	Function of various boiler mounting and accessories.	
		Introduction to modern boilers Benson boiler.	
	26 <sup>th</sup>	6. Properties of Steam:	
		Formation of steam and related terms. Thermodynamic	
		properties of steam,	
	27 <sup>th</sup>	Steam tables, sensible heat, latent heat, internal energy of	
		steam,	
$10^{th}$	$28^{th}$	entropy of water, entropy of steam, T- S diagrams,	
	29 <sup>th</sup>	2 <sup>nd</sup> sessional test (Tentative)	
	30 <sup>th</sup>	Assessment	
11 <sup>th</sup>	31 <sup>st</sup>	Mollier diagram (H – S Chart), Expansion of steam,	
· <del>-</del>	32 <sup>nd</sup>	Hyperbolic, reversible adiabatic and throttling processes,	
	32	Determination of quality of steam (dryness fraction),	
	33 <sup>rd</sup>	Unit V	
	33	7. Ideal and Real Gases	
		Concept of ideal gas, enthalpy and specific heat capacities of	
		an ideal gas,	
12 <sup>th</sup>	34 <sup>th</sup>	P - V - T surface of an ideal gas, triple point, real gases,	
14	34	Vander-Wall's equation.	
		vanuer-wan s equation.	

	35 <sup>th</sup>	8. Air Compressors	
	33		
		Functions of air compressor – uses of compressed air, type of	
	41-	air compressors	
	36 <sup>th</sup>	Single stage reciprocating air compressor, its construction and	
		working, Representation of processes involved on P – V	
		diagram, calculation of work done.	
13 <sup>th</sup>	37 <sup>th</sup>	Multistage compressors – advantages over single stage	
		compressors, use of air cooler,	
	38 <sup>th</sup>	condition of minimum work in two stage compressor (without	
	30	proof),	
	39 <sup>th</sup>	1 //	
	39	Rotary compressors – types, working and construction of	
, ,th	, o th	centrifugal compressor,	
14 <sup>th</sup>	40 <sup>th</sup>	axial flow compressor, vane type compressor	
	41 <sup>st</sup>	3 <sup>rd</sup> sessional test (Tentative)	
	42 <sup>nd</sup>	Assessment	
	12	ASSESSMENT	
15 <sup>th</sup>	43 <sup>rd</sup>	Revision	
13	43	Revision	
	, ,th		
	44 <sup>th</sup>	Revision	
	4 =th	5	
	45 <sup>th</sup>	Revision	

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Week	PRACTICAL		EXECUTION			
	Practical Day	Topic	G1	G2	Sign.	
1 <sup>st</sup>	Day 1 <sup>st</sup>	Introduction about the Lab and brief discussion over the practical work to be conducted.				
2 <sup>nd</sup>	2 <sup>nd</sup>	1.1 Determination of temperature by Thermocouple				
3 <sup>rd</sup>	3 <sup>rd</sup>	1.2 Determination of temperature by Pyrometer				
4 <sup>th</sup>	4 <sup>th</sup>	1.3 Determination of temperature by Infrared thermometer				
5 <sup>th</sup>	5 <sup>th</sup>	2. Study the working of Nestler boiler.				
6 <sup>th</sup>	6 <sup>th</sup>	Checking of Practical file/  1st sessional test (Tentative)				
7 <sup>th</sup>	7 <sup>th</sup>	3. Study of working of high pressure boiler.				
8 <sup>th</sup>	8 <sup>th</sup>	4.1 Demonstration of mountings on a boiler,				
9 <sup>th</sup>	9 <sup>th</sup>	4.2 Demonstration of accessories on a boiler				
10 <sup>th</sup>	10 <sup>th</sup>	Checking of Practical file/ 2nd sessional test (Tentative)				

11 <sup>th</sup>	11 <sup>th</sup>	5. Determination of Dryness fraction of steam using calorimeter.		
12 <sup>th</sup>	12 <sup>th</sup>	6. Demonstrate the working of air compressor.		
13 <sup>th</sup>	13 <sup>th</sup>	Checking of Practical file/ Evaluation		
14 <sup>th</sup>	14 <sup>th</sup>	3rd sessional test (Tentative)		
15 <sup>th</sup>	15 <sup>th</sup>	7. Study of boilers (Through industrial visit)		